Bicyclic Carbopalladation Reaction with Two *gem*-Reaction Centers. Efficient Construction of Fused Bicyclic Skeletons

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Synthetic efficiency is one of the most pursued objectives for chemists. Although bicyclic compounds are present in many important compounds, such as synthetic intermediates, target molecules of biological interests, etc.,^{1,2} among so many methodologies, one of the most commonly used approaches is a "ring by ring" strategy with low to moderate efficiency.^{3–7} We would like to propose here a new strategy for the efficient synthesis of bicyclic compounds, in which the two rings are expected to be formed *in one step* (Scheme 1). The advantage of this protocol is that *the bicyclic compounds with different ring sizes as well as functional groups can be prepared from the corresponding precursors in just "one shot"*.

Cyclic carbopalladation reaction of unsaturated C–C bonds has become one of the most powerful tools for the synthesis of cyclic compounds via a C–C bond-formation reaction,⁸ with overwhelming emerging of both new cyclization protocols and synthetic applications to complex target molecules. We began our hypothesis with a well-designed cyclic carbopalladation, i.e., *bicyclic carbopalladation reaction* of the corresponding precursors with two carbon–bromine bonds, in which the two bromine atoms are connected to the same carbon atom (Scheme 2). One formidable

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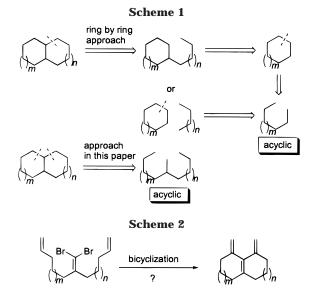
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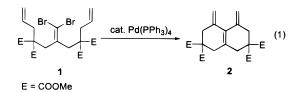
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challenge in this bicyclization is α -dehalopalladation,⁹ from which the formation of an sp²-carbon-centered carbene intermediate would be expected. We report herein the realization of our hypothesis. To our knowledge, this represents the first transitional metal-catalyzed bicyclic carbometalation reaction of *gem*-dibromides.¹⁰

We started our study with dibromide 1^{11} (eq 1). After



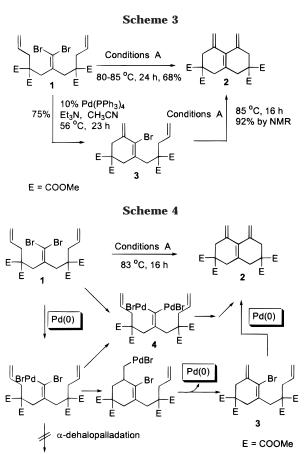
tedious attempts, finally we found that this reaction afforded fused 6,6-bicyclic product 2 in 68% yield in xylene at 80-85 °C for 24 h under the catalysis of 10 mol % Pd(PPh₃)₄ and $n-Bu_4NCl^{12}$ using K_2CO_3 as the base (conditions A). The results in CH₃CN, EtOH-DMF, and toluene were poor. Using Et₃N, Ag₂CO₃ (in CH₃CN), or K₂CO₃ (without n-Bu₄-NCl) as the base, the reactions were low-yielding and not clean. The ¹H NMR data are as follows: 5.23 (s, 2 H), 5.04 (s, 2 H), 3.70 (s, 12 H), 2.82 (s, 4 H), 2.63 (s, 4 H) ppm. The simplicity of the ¹H NMR data led us to doubt the structure of this cyclized product, since the six sp²-carbon atoms will not be on the same plane. This would, in principle, make the cyclic protons chemical environmentally unequal, thus complicating the splitting patterns. We determined the structure of the cyclization product as 2 unambiguously by X-ray study.¹³ α-Debromopalladation was not observed under the current reaction conditions.9

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⁽¹¹⁾ All the starting compounds were synthesized by the sequential alkylation of dimethyl malonate with 1,1,3-tribromo-2-(bromomethyl)propene and the corresponding alkenyl bromides or the reaction of allyl alcohol with 1,1,3-tribromo-2-(bromomethyl)propene in the presence of NaH in THF.



Under the catalysis of Pd(PPh₃)₄ and using CH₃CN as the solvent/Et₃N as the base, the monocyclization reaction of 1 to afford monocyclic product 3 was observed cleanly; even with longer reaction times the further cyclization of 3 was not observed. However, 3 cyclized to form 2 under conditions A in 92% NMR yield, indicating that 3 might be one of the reaction intermediates in the bicyclization shown in eq 1 (Scheme 3, for plausible mechanisms of this reaction, see Scheme 4).

colorless, a = 14.448(3) Å, b = 11.673(5) Å, c = 12.885(4) Å, 111.32(2) Å, $Z = 4, \mu$ (Mo K α) = 1.00 cm⁻¹, $D_{calc} = 1.287$ g/cm³, F(000) = 832.00, 2439 unique reflections were measured to $2\theta_{max} = 55.0$ °C, 1491 reflections with $I > 3.00\sigma$ -(1) were used in the refinement, R = 0.056.

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Under similar conditions, bicyclic ether 6 with one oxygen atom on each ring could be prepared in 83% yield (eq 2).

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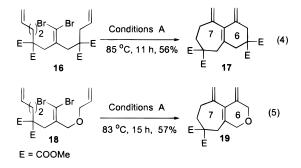
$$\begin{array}{c|c}
Br & Br \\
O & O \\
12 \\
\end{array} \\
\begin{array}{c}
Conditions A \\
85 \ ^{\circ}C, 15.5 \text{ h}, 83\% \\
13 \\
\end{array}$$
(2)

Although seven-membered ring formation via cyclic Heck reaction is not so well-documented,14 7,7-bicyclic compound **8** was prepared in 38% yield using $Ag_2CO_3^{15}$ instead of K_2 - CO_3 as the base (eq 3). The reaction under conditions A afforded 8 together with its C=C bond isomer(s).

$$E \xrightarrow{Br}_{E} E \xrightarrow{E}_{E} E \xrightarrow{Conditions A^{a}}_{E} \xrightarrow{T} \xrightarrow{T} \xrightarrow{T}_{E} E \xrightarrow{(3)}_{E}$$

^a Ag₂CO₃ (2.5 equiv) was used instead of K₂CO₃.

According to the strategy described in this paper, if two C=C double bonds are fixed with different side chains, the corresponding bicyclic compounds could be synthesized in one shot. Two such examples for the formations of 6,7bicyclic compounds 10 and 12 are listed in eqs 4 and 5.



In conclusion, we have developed the first bicyclic carbopalladation reaction, which was proven to be highly efficient for the construction of bicyclic compounds; α-dehalopalladation was not observed. The scope and cyclization patterns of these reactions are being investigated in our laboratory.

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Supporting Information Available: Experimental section, X-ray data of compound 2 and the ¹H NMR spectra of the compounds 2, 3, 6, 8, 10, and 12 (17 pages).

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⁽¹²⁾ For phase transfer-catalyzed intermolecular Heck reactions in DMF,